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Neutron, Proton, Electron are Questionable as the Basis of the Atom as Proton, Neutron, Electron Can be Seen as Reactions with Radium by British Scientists- We Can Return to A Mysterious Cloudy View of the Atom

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Abstract- Neutron discovered by James Chadwick could be questioned as alpha causes particle emission reaction product. Neutrons may not be fundamental but rather a reaction product. Electron discovered by JJ Thompson could be questioned as particles deflect to positive plate are possibly a gas or gravity effect? Proton could be questioned as radium effect with Au or Gold not fundamental.

Keywords: *electron, proton, neutron, radium, fundamental particles.*

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I. Introduction

The discovery of fundamental particles neutron, electron and proton was based on reactions with radium by a few British scientists Ernest Rutherford, James Chadwick and JJ Thompson soon after the discovery of radium. One can argue all three fundamental particles are reaction products of radiation with radioactive radium and not really fundamental particles.

II. Methodology

3 Basic fundamental particles proton, electron, neutron are questionable based on the origin or discover of electron, proton and neutron. One could say electron, proton and neutron are reaction products with radium and not fundamental particles. Much like we see light as wave particle reality; we can argue again that the atom also is partly a reaction product with radium and really an electron, proton and neutron. [1]

III. Results and Discussion

We can return to the idea of a somewhat mysterious atom as all three fundamental particles could or may be reaction products with radioactive radium. Radium chemistry and radium physics can be seen as the origin of the electron and proton and neutron not really fundamental particles. Electron, proton, neutron may have been reaction products with radium and not fundamental parts of the atom.

IV. Conclusion

Radium reactions are the basis of the British discovered electron, proton and neutron. James Chadwick threw radium at atoms and got radium reactions. Rutherford threw radium at gold and got radium reactions not fundamental particles. JJ Thompson used a Crookes and possibly learned about the way radium behaves not the fundamental particles. British discoveries of fundamental particles are limited in importance and the whole atom may actually be more of a mysterious atom.

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References Références Referencias

1. References to how the Neutron, Proton, Electron were discovered or accessed from yahoo and google website searches on these fundamental particles origins.

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